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*Organic Chemistry 1 | Final
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Review - List of Reactions
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~~Review Session | Monday,
November 30th How to~~

~~Memorize Organic Chemistry
Reactions and Reagents~~

~~[Workshop Recording] ACS~~

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~~Final Exam (Nov. 26, 2019)~~

ACS Organic Chemistry I Exam

1 Review | Monday, September

21st *Organic Chemistry 1*

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Guide Multiple Choice Test

Youtube ACS Organic

Chemistry II Exam II Review

(March 17, 2020)

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SN1/SN2/E1/E2 Mechanisms Do
not be afraid of organic
chemistry. | Jakob Magolan |
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How To Get an A in Organic
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2018** *Tips \u0026 Tricks for
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Study Guide - IB, AP, \u0026
College Chem Final Exam ACS
Organic Chemistry Final Exam
Review - Acids and Bases ACS
EXAM REVIEW 2 ACS Organic
Chemistry Review Exam 3
(Nov. 12, 2019) ACS Exam
Tips for Chem Students: How
to Take the ACS Exam ACS
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Review - Spectroscopy ACS
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Session | Exam 2 | Tuesday,
October 27th ~~Acs Organic
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ACS Examination guide
(Selected Questions) Organic
Chemistry Nomenclature 1.
What is the IUPAC names for
this compound? a) 1-tert-
butyl-2-butanol b)*

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5,5-dimethyl-3-hexanol c)

2,2-dimethyl-4-hexanol d)

1,1,1-trimethyl-3-pentanol

4. What is the IUPAC name for this structure? a)

3-bromo-4-methylheptanone b)

5-bromo-4-methylheptanone

~~ACS Examination guide~~

~~(Selected Questions) Organic~~

~~...~~

Topics Covered.

Nomenclature; Structure,

Hybridization, Resonance,

Aromaticity; Acids and

Bases; Stereoisomerism;

Nucleophilic Substitutions

and Eliminations

~~Organic Chemistry | ACS~~

~~Exams~~

Preparing for Your ACS

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r Organic Chemistry Standard
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Organic chemistry is the
study of the structure,

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~~Solutions~~, composition, reactions, and preparation of carbon-containing compounds, which include not only hydrocarbons but also compounds with any number of other elements, including hydrogen (most compounds contain at least one carbon-hydrogen bond), nitrogen, oxygen, halogens, phosphorus, silicon, and sulfur.

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Page I-60 / CH 223 A.C.S.

Final Exam Study Guide.

American Chemical Society
(A.C.S.) General Chemistry

Exam. 75 multiple choice
questions, 110 minutes Bring
a scantron, calculator and
pencil(s) I. States of

Matter. Ideal Gas Law: $PV = nRT$, $PM = dRT$, deviations
from ideal (high P, low T)

Ionic Solids: sc, fcc, bcc

Phase Diagrams: Pressure
versus Temperature

Intermolecular Forces: ion-
dipole, hydrogen bonding,
etc.

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stars 7

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of sample exam or anything
close relating to their
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