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An electron is contained in a three-dimensional rectangular box given by $0 \leq x \leq a$, $0 \leq y \leq b$, and $0 \leq z \leq c$. The solutions of the Schrödinger equation are specified by the quantum numbers n_x , n_y , and n_z . Recall that the normalized wave function is...with energy...where n_x , n_y , and n_z are positive integers.

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$l(l+1) \hbar^2 m^{-2} = 1/4 \hbar^2 m^{-2}$, which is to say,
 $l^2 + l + 1/4 = \hbar^2 m^{-2}$, or $(l + 1/2)^2 = \hbar^2 m^{-2}$, or $l =$
 $|\hbar m| \hbar^{-1} = 1/2$. [Our present equation depends only
on $\hbar^2 m^{-2}$, and hence is the same for either

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sign, but the solution to Problem 4.39 assumed $l+1/2 \neq 0$ (else u is not normalizable), so we need $|m|$ here.] Quoting 4.39: $E = (j$.

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